Superstrong supercooled zeolitic hybrid frameworks and topological melting
Greaves, N.; Tan, J.C.; Yue, Yuanzheng; Bennett, T.; Cheetham, A.; Zhou, Z.F.

Publication date:
2014

Document Version
Accepted author manuscript, peer reviewed version

Link to publication from Aalborg University

Citation for published version (APA):
Polyamorphism is a supercooled phenomenon, resulting in glassy phases having the same composition, but differing in entropy and density. These also occur during amorphisation, not least for zeolites which readily collapse to form novel glasses under modest pressures or when temperatures approach the glass transition \( T_g \). A sequence of phase transitions is involved: the first, topologically invariant, converting the microporous crystal into a low density low entropy phase; the second, topologically disordering, transforming this into a polymorph, similar what is recovered by melt-quenching (Greaves et al, *Science* 308, 1219 (2005)). Moreover, the melting temperature \( T_m \) can be projected from \( T_g \) from the well-known empirical relation \( T_m \sim 3/2T_g \). It coincides for inorganic zeolites with feldspar melting points. More recently amorphisation of zeolitic imidazole frameworks (ZIFs) has been demonstrated leading to hybrid glasses (Bennett et al, *PRL* 104, 115503 (2010)). These are microporous (Chapman et al, *JACS* 131, 17546 (2009)) and also mechanically stronger than their ZIF precursors (Tan & Cheetham, *Chem. Soc. Rev.* 40, 1059 (2011)). We report in situ X-ray scattering and DSC experiments of ZIF 4 amorphised at different heating rates from which we observe the glass polyamorphic transition temperatures \( T_g \) and the associated rheological properties. As with inorganic zeolites, we find that the supercooled ZIF liquid is extremely strong and reminiscent of low entropy low density polymorphs. Furthermore evidence for an order-disorder transition resides in dramatic peaks in X-ray scattering with rising temperature. These reflect deep minima in the bulk modulus at the liquid-liquid transition, typical of first order transitions (Greaves et al, *Nat. Mat.* 10, 823 (2011)). Finally for zeolitic hybrids we define a topological melting point \(~3/2T_g\), sometimes unreachable because of the dissociation of the organic imidazole linkages at intermediate temperatures, but affording a measure of the mechanical stability of the hybrid structure.