

Aalborg Universitet

Characterization of Phase Separation Processes in modifier-free binary Al2O3-SiO2 glasses by Solid State NMR

Mönster, C.; Eckert, H.; Yue, Yuanzheng

Publication date: 2006

Document Version Accepted author manuscript, peer reviewed version

Link to publication from Aalborg University

Citation for published version (APA):

Mönster, C., Eckert, H., & Yue, Y. (2006). Characterization of Phase Separation Processes in modifier-free binary Al2O3-SiO2 glasses by Solid State NMR. Poster presented at 27th Danish NMR discussion meeting, Rebild, Denmark.

Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

- Users may download and print one copy of any publication from the public portal for the purpose of private study or research.
- You may not further distribute the material or use it for any profit-making activity or commercial gain
 You may freely distribute the URL identifying the publication in the public portal -

Take down policy
If you believe that this document breaches copyright please contact us at vbn@aub.aau.dk providing details, and we will remove access to the work immediately and investigate your claim.

Characterization of Phase Separation Processes in modifier-free binary Al₂O₃-SiO₂ glasses by Solid State NMR



Christine Mönster, Hellmut Eckert, Yuanzheng Yue
Westfälische Wilhelms-Universität Münster
Institut für Physikalische Chemie and NRW Graduate School of Chemistry
Corrensstr. 30, 48149 Münster, Germany
monster@uni-muenster.de



Introduction

Many disordered materials used in technological applications are phase separated on the micro- or nanoscale. Prominent examples are bioactive glasses and glass ceramics which have applications in restorative surgery and as body implants. To optimize the chemical and mechanical properties of these biomaterials in an efficient manner, controlled crystallization of the initial glass ceramic composition can be induced by thermal treatment of glasses. We are interested in questions concerning structural ordering phenomena in the amorphous matrix before the crystallization takes place. To this end we chose $0.35 \, \text{Al}_2 \text{O}_3 \,^* \, 0.65 \, \text{SiO}_2$ as a model system.

The structure of non-crystalline materials in the system Al_2O_3 -SiO₂ has been a point of interest for many years [1,2]. The glass forming ability of silica melts with considerable amounts of alumina is low, but roller-quenching techniques and sol-gel methodology have made a preparation possible [3,4]. We are interested in the comparability of the micro-structures of these with different techniques obtained glasses. In a second step the base glasses were annealed at different temperatures below and above the glass transition.

Synthesis of 0.35 Al₂O₃ * 0.65 SiO₂ ...

... with Hyperquenching Techniques

The samples of the alumosilicate glass ceramic system have been provided by Dr. Y. Yue from Aalborg University.

AIO₄

973°C

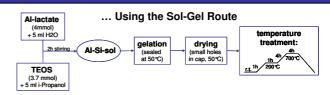
875°C

760°C

125

AIO,





Parts of each base glass were annealed for 90 min. at different temperatures: 760 ℃, 875 ℃, 973 ℃ and 1075 ℃.

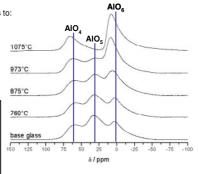
²⁷AI-NMR ...

... of the Hyperquenched Samples

- Annealing up to 875 ℃:
- constant AlO₄ content
 constant AlO₅ content
- > constant AlO₅ content
- Annealing at 973°C: disappearance of AlO₅; crystallization to mullite

... of the Sol-Gel Derived Samples

- · Rising annealing temperature leads to:
- ➤ increasing AIO₆ content
- >decreasing AlO₅ content >constant AlO₄ content
- ➤ shift of AlO₆ to higher frequency
- •Annealing at 1075 °C: disappearance of AlO₅; crystallization to mullite ??
- The spectra of the hyperquenched samples show broader lines and not such a clear separation between the different groups like in the sol-gel derived samples.
- In the sol-gel samples the disappearance of the AlO₅ environment is a continuous process that is already visible at an annealing temperature of 875°C and is completed at 1075°C, whereas in the hyperquenched samples this seems to happen suddenly between 875°C and 973°C.



²⁹Si-NMR ...

... of the Hyperquenched Samples

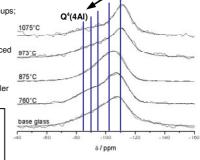
973°C Q4(4AI) 973°C 875°C 760°C base glass 120 120 140 160

å/ppm

- Base glass: consists mainly of Q⁴(1Al) groups
- Increasing annealing temperature:
- first formation of a shoulder on the left side
- > finally separation into two peaks
- Annealing at 973 ℃: large amount of Q⁴(0Al) and smaller one of Q⁴(4Al)

... of the Sol-Gel Derived Samples

- Base glass: consists of a distribution of Q⁴ groups; Q⁴(0Al) up to Q⁴(4Al)
- Increasing annealing temperature:
- Increasing annealing temperature:
 shoulder becomes more pronounced
- ➤ amount of Q⁴(0Al) increases
- Annealing at 1075 °C: large amount of Q⁴(0AI) and smaller one of Q⁴(2AI) up to Q⁴(4AI)



Q⁴(QAI)

- As well in the hyperquenched as in the sol-gel derived samples a phase separation into silicon rich (Q4(0Al)) and aluminum rich (Q4(3 or 4 Al)) areas takes place with increasing annealing temperature.
- This process is in the hyperquenched system strongly pronounced, whereas it seems not yet completed in the sol-gel system when annealing at 1075 °C.

Conclusions

- Alumosilicate glasses (0.35 Al₂O₃* 0.65 SiO₂) obtained with hyperquenching techniques and via the sol-gel route were proven to have similar silicon and aluminum environments.
- Annealing the base glass at different temperatures below and above the phase transition leads in the sol-gel derived glasses to increasing AlO_6 environment and decreasing AlO_5 environment. Annealing at $1075\,^{\circ}\text{C}$ vanishes all five coordinated aluminum. In the hyperquenched system the AlO_5 environment disappears suddenly between $875\,^{\circ}\text{C}$ and $973\,^{\circ}\text{C}$.
- 29Si-NMR shows phase separation into silicon rich and aluminum rich areas with increasing annealing temperature. This process is more pronounced in the hyperquenched system and seems to need higher annealing temperatures in the sol-gel system.

Literature

[1] Morikawa, H. et al.: J. Am. Ceram. Soc. **65** (1982) 78 [2] Risbud, S. et al.: J. Am. Ceram. Soc. **70** (1997) C10 [3] Schmücker, M. et al.: J. Non-Cryst. Solids **217** (1997) 99 [4] Eckert, H.: Prog. NMR Spectrosc. **24** (1992) 159

Acknowledgements

C.M. thanks The NRW Graduate School of Chemistry for a