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## On the biosynthetic origin of carminic acid

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### ABSTRACT

The chemical composition of the scale insect *Dactylopius coccus* was analyzed with the aim to discover new possible intermediates in the biosynthesis of carminic acid. UPLC-DAD/HRMS analyses of fresh and dried insects resulted in the identification of three novel carminic acid analogues and the verification of several previously described intermediates. Structural elucidation revealed that the three novel compounds were desoxyerythrolaccin-*O*-glucosyl (**DE-O-Glc**), 5,6-didehydroxyerythrolaccin 3-*O*-β-D-glucopyranoside (**DDE-3-O-Glc**), and flavokermesic acid anthrone (**FKA**). The finding of **FKA** in *D. coccus* provides solid evidence of a polyketide, rather than a shikimate, origin of coccid pigments. Based on the newly identified compounds, we present a detailed biosynthetic scheme that accounts for the formation of carminic acid (**CA**) in *D. coccus* and all described coccid pigments which share a flavokermesic acid (**FK**) core. Detection of coccid pigment intermediates in members of the *Planococcus* (mealybugs) and *Pseudaulacaspis* genera shows that the ability to form these pigments is taxonomically more widely spread than previously documented. The shared core-FK-biosynthetic pathway and wider taxonomic distribution suggests a common evolutionary origin for the trait in all coccid dye producing insect species.

### 1. Introduction

Pigments derived from insects and especially coccids (scale insects) have been used by humans since ancient times for dyeing textiles, in cosmetics and in paints, and for coloring foods (Donkin, 1977). The most commonly used coccid dyes include kermesic acid (**KA**), laccaic acids (**LA**) and carminic acid (**CA**), which share a red color hue due to a similar chromophore structure (Łagowska and Golan, 2009). The compounds, or combinations of these, have been reported to be produced by several species of distantly related scale insects (Hemiptera: Coccoidea). Mainly five species, namely *Porphyrophora hamelii* (Armenian/Ararat cochineal), *Kermes vermilio* (kermes), *Porphyrophora polonica* (Polish cochineal), *Dactylopius coccus* (Mexican cochineal) and

*Kerria lacca* (Indian lac insect) have at various points in history, and at different geographical localities, been utilized by humans for large scale production of coccid dyes (Donkin, 1977). Carminic acid and its aluminum salt carmine (E120) is by many considered as the pinnacle of coccid dyes, based on its hue, light, temperature, and oxidation stability, and the yields by which it can be obtained from natural sources (Dapson, 2007). **CA** is known to be produced by *P. hamelii* (Asia Minor), *P. polonica* (Europe), and *D. coccus* (Meso and South America), all of which have served as sources for the compound (Wouters and Verheken, 1989). Present day production is based on *D. coccus* due to its exceptional high pigment content (16–22% of dry weight), low fat content, and the ease by which the insect can be cultured and harvested from cladodes (leaves) of *Opuntia* cacti (Donkin, 1977; Downham and

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Collins, 2000). A thorough introduction to the historical use and geopolitical role of carmine is given by Dapson (2007).

Although insect-derived pigments have been utilized by humans for millennia and remain of significant value within the food colorant market, the underlying biochemistry for their production remains largely unknown. The coccid dyes, such as CA, have by many authors been categorized as polyketides solely based on their structure (Morgan, 2010; Cameron et al., 1978; Pankewitz and Hilker, 2008; Brown, 1975). The biosynthetic mechanisms of formation of polyketides, via the successive condensation of acetyl-CoA and malonyl-CoA units catalyzed by polyketide synthases (PKSs), is well described in bacteria, fungi, and plants (Staunton and Weissman, 2001). However, no animal PKSs have yet been biochemically characterized, even though many insect species are known to contain compounds that potentially may be synthesized via the polyketide pathway. In most cases, the putative polyketides contents have been ascribed to the sequestering of precursors, or the finished compounds, from the insects' diet (Pankewitz and Hilker, 2008). This situation is seen in *Timarcha* spp. (leaf beetles) which accumulate anthraquinones from its host plant *Galium* spp. (Rubiaceae) (Petitpierre, 1981), and in *Laetilia coccidivora* (pyralid moth) and *Hyperaspis trifurcata* (coccinellid beetle) larva that accumulate CA by preying on *Dactylopius* spp. (Eisner et al., 1994). In other cases, the origins of the detected polyketides in insects have been linked to the activity of endosymbiotic bacteria, such as the production of pederin, a polyketide-peptide hybrid, in *Paederus* spp. (rove beetles), which depends on an endosymbiotic bacterium related to *Pseudomonas aeruginosa* (Piel, 2002; Kellner, 2002). A second example is the facultative endosymbiotic *Rickettsiella* spp. responsible for the production of the polyketide viridaphin A1 in various aphids (*Acyrtosiphon pisum* and *Megoura crassicauda*) (Tsuchida et al., 2010; Horikawa et al., 2011). The biosynthetic origin of coccid pigments in scale insects, however, remains a mystery. The hypothesis that coccid dyes are polyketides has solely been based on their chemical structure, even though the polyketide class is characterized by a shared mode of synthesis rather than shared structural features (Staunton and Weissman, 2001). In fact, for the anthraquinone core of CA, one could just as well envision that this is formed via the shikimate based chorismate/*O*-succinyl benzoic acid pathway, as described for lucidi, alizarin, and morindone in rubiaceous plants (Leistner, 1973; Han et al., 2001). If CA is formed by this pathway it would not qualify as a polyketide and its formation would not be dependent on a PKS. However, it is possible to distinguish between the two alternative pathways as the polyketide-based synthesis would include a unique anthrone intermediate, which is not found in the shikimate-based pathway, where the anthraquinone is formed directly.

The present study aims at increasing our understanding of the CA origin and its biosynthesis in *D. coccus*. We also aim to elaborate on the previously proposed links to other biosynthetic pathways responsible for the production of coccid dyes within the *Coccoidea* superfamily. In our study, we report the presence of flavokermesic acid anthrone (FKA) in *D. coccus*, which strengthens the hypothesis that coccid pigments are formed via the polyketide pathway, as FKA is the first cyclic intermediate in a polyketide-based pathway.

## 2. Materials and methods

### 2.1. Biological material

Adult *D. coccus* specimens were collected from *Opuntia* cacti pads on the Canary Islands, Lanzarote, near the village of Guatiza, June 2012. The insects were transported to Denmark either as live specimens on cacti pads or as dead specimens stored on dry ice. Additional *D. coccus* insects were collected from cacti pads near the city of Arequipa Peru, August 2012, flash frozen in liquid nitrogen, and shipped to Denmark on dry ice. Commercially available dried *D. coccus* insects were supplied by Chr. Hansen A/S.

*Coccus hesperidum*, *Pseudococcus longispinus*, *Palmicultor browni*, and *Pseudaulacaspis pentagona* were collected in the greenhouses of the Botanical Garden (Natural History Museum of Denmark, University of Copenhagen) in Copenhagen in June 2014, and identified using the latest available identification keys (Dooley and Dones, 2015; Miller et al., 2014).

### 2.2. Instrumentation

Chemical analysis of *D. coccus* samples was performed using three different LC-MS setups. UPHLC-DAD-HRMS was performed on a maxis G3 QTOF mass spectrometer (Bruker Daltonics, Bremen, Germany) equipped with an electrospray ionization source coupled to an Ultima 3000 UHPLC-DAD (Dionex). Separation was performed on a Kinetex C<sub>18</sub> column (150 × 2.1 mm, 2.6 μm, Phenomenex Inc., Torrance, CA, USA) maintained at 40 °C using a linear H<sub>2</sub>O-acetonitrile gradient consisting of A: milliQ H<sub>2</sub>O containing 10 mM formic acid and B: acetonitrile containing 10 mM formic acid from 10 to 100% B in 10 min with a flow rate of 400 μL min<sup>-1</sup>. The FK anthrone was detected on a HPLC-DAD-HRMS system consisting of an Agilent 1200 chromatograph comprising quaternary pump, degasser, thermostatted column compartment, autosampler, and photodiode array detector (Agilent Technology, Santa Clara, CA, USA) and a Bruker micrOTOF-Q II mass spectrometer (Bruker Daltonik, Bremen, Germany) equipped with an electrospray ionization source and operated via a 1:99 flow splitter. Analyses were performed at 40 °C on a Luna C<sub>18</sub>(2) reversed-phase column (150 × 4.6 mm, 3 μm particle size, 100 Å pore size, Phenomenex Inc., Torrance, CA, USA) with a flow rate of 800 μL min<sup>-1</sup>. HPLC solvent A consisted of H<sub>2</sub>O-acetonitrile 95:5 (v/v) with 0.1% formic acid and solvent B consisted of acetonitrile-H<sub>2</sub>O 95:5 (v/v) with 0.1% formic acid. Separation was obtained using a linear gradient from 0 to 100% B in 20 min. Mass spectra were acquired in negative ionization mode. The search for coccid dye intermediates in the different scale insect species was performed on a 6540 Ultra High Definition UHD Accurate Mass Quadrupole Q-TOF LC/MS system (Agilent Technology, Santa Clara, CA, USA). Separation of the analytes was conducted on a Kinetex XB-C<sub>18</sub> (100 × 4.6 mm i.d. 2.6 μm, Phenomenex Inc., Torrance, CA, USA), column maintained at 35 °C. The analytes were eluted using a H<sub>2</sub>O-acetonitrile gradient consisting of A: MilliQ H<sub>2</sub>O containing 10 mM formic acid and B: acetonitrile containing 10 mM formic acid starting with 50 s, 0% B; and then a gradual increase 110 s, 18.6% B; 170 s, 37.8% B; 290 s, 52.2% B; 360 s, 54.2% B; 480 s, 90% B; 600 s, 100% B; 660 s, 100% B with a flow rate of 400 μL min<sup>-1</sup>. The column was reconstituted with 100% A for 110 s prior to injection of the subsequent sample.

NMR spectra of 5,6-didehydroxyerythrolaccin 3-*O*-β-D-glucopyranoside (DDE-3-*O*-Glc), desoxyerythrolaccin *O*-glucopyranoside (DE-*O*-Glc), and dcII were recorded on a Varian Unity Inova 500 MHz (Varian Inc., Palo Alto, California) using a 5-mm probe. Samples were dissolved in 500 μL DMSO-*d*<sub>6</sub> and referenced to δ<sub>H</sub> at 2.50 ppm and δ<sub>C</sub> at 39.5 ppm. The NMR spectrum of flavokermesic acid anthrone (FKA) was recorded on a Bruker Avance III HD 600 MHz NMR spectrometer (<sup>1</sup>H operating frequency 600.13 MHz) equipped with a cryogenically cooled 5-mm DCH probe (Bruker Biospin, Rheinstetten, Germany). The sample was dissolved in acetone-*d*<sub>6</sub> and referenced to δ<sub>H</sub> 2.05 ppm and δ<sub>C</sub> 29.84 ppm. Following structural elucidation of the described compounds, their presence in the original samples was verified using targeted MS analysis.

Chiral GC-MS was performed using 10 μg DDE-3-*O*-Glc that was hydrolyzed in 10% aqueous HCl for 90 min at 90 °C, dried by a stream of N<sub>2</sub> and dissolved in 40 μL dry pyridine followed by 10 μL *N*-methyl-bis-trifluoroacetamide (MBTFA) (GC-grade, 99%, Sigma-Aldrich) and heating to 65 °C for 40 min. The sample was cooled to room temperature and subsequently analyzed on a CP-ChiraSil-L-Val GC column (25 m × 0.25 mm × 0.12 μm, Agilent Technology, Santa Clara, CA, USA) programmed to 70–150 °C at 4 °C min<sup>-1</sup> using an HP 6890 series

GC system and Agilent 5973 mass selective detector. The sample was compared to the standards of D- and L-glucose (Sigma-Aldrich, St. Louis, MO, USA).

### 2.3. Extraction and isolation

For analysis of metabolites in fresh insects approximately 1 g of flash frozen material was grinded and extracted with MeOH:H<sub>2</sub>O (1:1). Desoxyerythrolaccin (**DE**) and **DDE-3-O-Glcp** were purified from 100 g of dried *D. coccus*. The insects were grinded and extracted with MeOH:H<sub>2</sub>O (1:1) followed by liquid-liquid partitioning with EtOAc at pH 3 and concentrated *in vacuo*. The crude extract was first separated by ion-exchange using an NH<sub>2</sub> flash column. Carboxylic acid containing compounds was retained when washed with 50% aqueous MeCN containing 10 mM ammonium formate. Finally, the acidic compounds (**FK**, **KA**, and **CA**) were eluted with 50% aqueous MeCN adjusted to pH 11 with ammonium hydroxide. **DE** and **DDE-3-O-Glcp** were purified from the 50% MeCN 10 mM ammonium formate eluate using reversed-phase semi-preparative chromatography on a 250 × 10 mm Luna C<sub>18</sub>(2) column (Phenomenex, Torrance, CA, USA) using a Gilson HPLC system. Compounds were eluted with a gradient consisting of MilliQ H<sub>2</sub>O:MeCN, both containing 50 ppm TFA. **FK** and **dcII** were recovered in the alkaline eluent of the NH<sub>2</sub> column. The extract was then further purified on Isolute diol material (Biotage, Uppsala, Sweden) on an Isolera auto flash purification system (Biotage, Uppsala, Sweden) in a step-wise elution from dichloromethane to EtOAc to MeOH to afford **dcII** and **FK**.

### 2.4. Synthesis of flavokermesic acid anthrone

Synthesis of flavokermesic acid anthrone from flavokermesic acid was conducted according to a previously published method by Schätzle, with slight modifications (Schätzle et al., 2012). In brief, 10 mg flavokermesic acid was dissolved in 1 mL glacial acetic acid and 0.2 mL hydriodic acid (57 wt % in H<sub>2</sub>O) in a sealed microwave reactions vial and heated to 50 °C under stirring for 2 h in the dark. This adaption to the method reported by Schätzle et al., 2012) was done to ensure full conversion of the flavokermesic acid to the anthrone with only limited decarboxylation. The hydriodic acid was quenched with 10 mL saturated Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and extracted three times with diethyl ether. The ether phase was dried over MgSO<sub>4</sub> and lyophilized under reduced pressure. The sample was at all times kept in the dark and chemical analyses were performed in amber vials and NMR tubes to minimize the risk of dimerization and oxidation. The formation of the anthrone was confirmed by NMR analysis (Supplementary data Table S1).

### 2.5. Biosynthetic models

The biosynthetic models for formation of **CA** and related compounds were formulated using the retrosynthetic analysis approach including commonly accepted enzymatic driven reactions, as described in the BRENDA database (Schomburg et al., 2004), and the available structural data for coccid dyes (Morgan, 2010; Cameron et al., 1978, 1981; Brown, 1975; Peggie et al., 2007; Stathopoulou et al., 2013; Bhide et al., 1969). The models were drawn using ChemDraw 15.9.9.106 (PerkinElmer Informatics, Inc., US).

## 3. Results

### 3.1. Detection of compounds in *Dactylopius coccus* extracts

HPLC-HRMS analysis of the raw extracts from fresh *D. coccus* showed that the main extractable pigment components were **CA**, **FK**, **KA** and **dcII** as previously reported (Wouters and Verheken, 1989; Peggie et al., 2007; Méndez et al., 2004) (Fig. 1A). These compounds were putatively identified based on relative retention time, high-

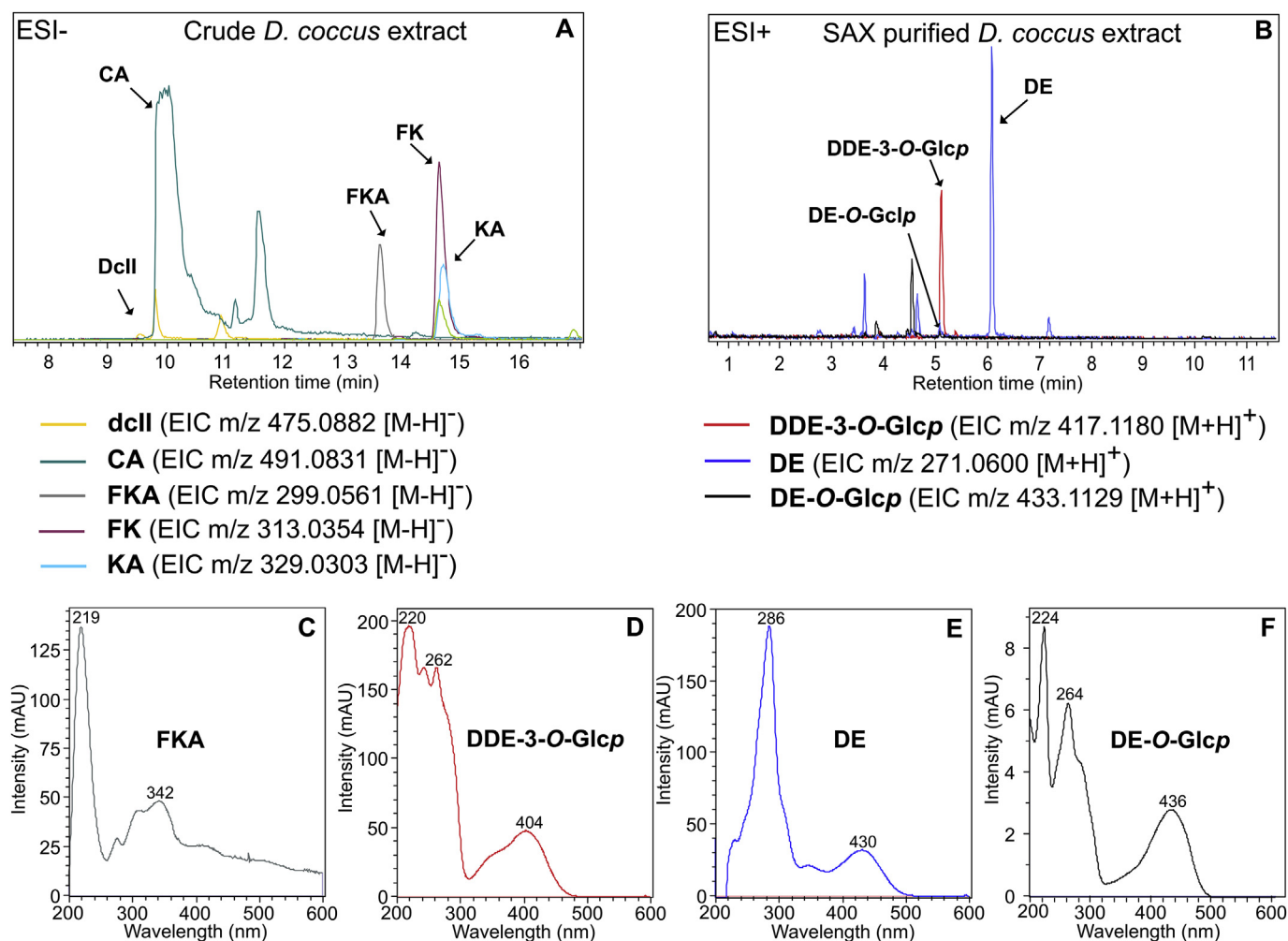
resolution mass, UV/VIS spectra, and MS/MS fragmentation patterns. Spiking with authentic samples of **CA** and **KA** confirmed these compounds identity. **FK** was isolated and subjected to structural elucidation by 2D NMR (Supplementary data Table S1), and the data were in agreement with and confirmed the previously reported structure of **FK** (Wouters and Verheken, 1987). **CA** and **dcII** were not easily separated using reversed-phase HPLC, and isolation of the two compounds relied on normal-phase flash chromatography on diol substituted silica. The putative **dcII** was further purified using semi-preparative HPLC and characterized by HR-MS and 2D NMR. Structural elucidation of the compound by 2D NMR experiments (Supplementary data Table S1) showed that **dcII** was flavokermesic acid 2-C-β-D-glucopyranoside, and the NMR data were in agreement with those reported for **dcII** (Stathopoulou et al., 2013). The reversed-phase HPLC-based analysis also revealed a previously undescribed major peak, eluting at 13.40 min (Fig. 1A and C). The mass of the corresponding compound equaled the theoretical mass of flavokermesic acid anthrone (**FKA**), and its identity was confirmed by comparison with a **FKA** standard, semi-synthesized from authentic **FK**. In addition, to the metabolites detected using reversed-phase chromatography, strong anion exchange SPE (SAX SPE) were used to identify three non-acidic metabolites (**DE**, **DDE-3-O-Glcp** and **DE-O-Glcp**) that all displayed UV/VIS spectra with similarities to those reported for **FK**, **KA**, **dcII** and **CA** (Fig. 1B, D, 1E and 1F). For unambiguously structure elucidation of **DE** and **DDE-3-O-Glcp**, the compounds were purified in amounts sufficient for structural elucidation by HRMS and NMR spectroscopy. Purification of the **DE-O-Glcp** compound unfortunately did not yield sufficient quantities for full structural elucidation by NMR.

### 3.2. Structural elucidation of **DE**, and the novel compounds **DE-O-glucosyl**, **DDE-O-glucosyl** and **FKA**

**DE** displayed UV/VIS spectrum (Fig. 1E) similar to that of **FK**, indicating a similar core skeleton. The compound was not retained on a SAX column, suggesting that it lacked the carboxylic acid group found at C-7 in **FK**. This conclusion was supported by HRMS ( $m/z$  271.0600  $[M+H]^+$ , calcd 271.0600,  $\Delta M$  0.0 ppm), suggesting a molecular formula of C<sub>15</sub>H<sub>10</sub>O<sub>5</sub>, i.e., **DE** lacking CO<sub>2</sub> as compared to **FK**. Structural elucidation was carried out by <sup>1</sup>H NMR and 2D NMR spectroscopy (Table S1). The <sup>1</sup>H NMR spectrum showed a signal for the OH-group positioned *peri* to the carbonyl group ( $\delta$  13.30, 1-OH), two sets of meta-coupled protons H-5 and H-7 ( $\delta$  6.54 and 7.04, respectively, <sup>3</sup>J<sub>H5-H7</sub> = 1.9 Hz) and H-2 and H-4 ( $\delta$  7.43 and 7.01, respectively, <sup>3</sup>J<sub>H2-H4</sub> = 2.5 Hz), and a *peri*-positioned methyl group ( $\delta$  2.81, s, 11-CH<sub>3</sub>). The meta-coupling between H-5 and H-7 clearly proves the lack of the carboxylic acid in position 7, and thus HRMS and NMR data supported the compound to be **DE**, also known as 3-hydroxy-aloesaponarin II (Mehandale et al., 1968), and previously observed in air-dried *D. coccus* (Sugimoto et al., 1998).

Structural analysis of the first novel compound **DDE-3-O-Glc** revealed that the compound was a hitherto undescribed O-glucoside of 5,6-didehydroxyerythrolaccin (**DDE**) (Fig. 2). The high-resolution mass spectrum of **DDE-3-O-Glcp** suggested a molecular formula of C<sub>21</sub>H<sub>20</sub>O<sub>9</sub> ( $[M+H]^+$   $m/z$  417.1180, calcd. C<sub>21</sub>H<sub>21</sub>O<sub>9</sub><sup>+</sup> 417.1180,  $\Delta M$  0.0 ppm;  $[M-H]^-$   $m/z$  415.1034, calcd. C<sub>21</sub>H<sub>19</sub>O<sub>9</sub><sup>-</sup> 415.1029,  $\Delta M$  1.2 ppm). In addition, the compound exhibited a loss of  $m/z$  162.0528, which is likely due to the loss of a labile O-linked hexose unit. The structural elucidation was carried out based on <sup>1</sup>H NMR and 2D NMR spectroscopy (Supplementary data Table S1). The <sup>1</sup>H spectrum showed resemblance to that of **DE**, but instead of the two doublets observed for the meta-coupled H-5 and H-7 in **DE**, signals for H-5 ( $\delta$  8.12, dd, 7.5, 1.2 Hz), H-6 ( $\delta$  7.78, t, 7.5 Hz), and H-7 ( $\delta$  7.75 dd, 7.5, 1.3 Hz) showed the absence of a hydroxyl group at C-6 in **DDE-3-O-Glcp**. Furthermore, a doublet at  $\delta$  5.26 (<sup>3</sup>J<sub>H1'-H2'</sub> = 7.6 Hz) for a β-configuration of the anomeric proton H-1' (as well as the <sup>13</sup>C value of 101.1 ppm for C-1) and the remaining <sup>1</sup>H and <sup>13</sup>C signals for H-2' to H-6' and C-2' to C-6'





**Fig. 1.** HPLC-HRMS analysis of CA related compounds found in *Dactylopius coccus* with annotation of major peaks and UV/VIS spectra for the novel compounds. **A:** Annotated EIC chromatogram for the CA related compounds identified in the raw extract of *D. coccus*, coloring scheme for the EIC is shown below the chromatogram. **B:** Annotated EIC chromatogram for CA related compounds identified in the SAX purified *D. coccus* extract, **C-F:** UV/Vis spectra for DE and the new compounds DDE-3-O-Glcp, DE-O-Glcp, and FKA.

(Supplementary data Table S1), are in agreement with a  $\beta$ -D-glucose unit (Bock and Pedersen, 1983). The O-linkage of the sugar was evident from a more deshielded anomeric proton (101.1 ppm) compared to that of C-glucosyl linkages reported for related compounds (Stathopoulou et al., 2013). In addition, a HMBC correlation from H-1' to C-3 (164.6 ppm) further confirmed the O-glucosyl linkage to C-3. Thus, to establish the D- or L-configuration of the glucose moiety, an aliquot was hydrolyzed and analyzed by GC-MS and chiral GC-MS. This confirmed that the hexose moiety was D-glucose (Figure S2 and S3). Thus, the compound was identified as 5,6-didehydroxyerythrolaccin 3-O- $\beta$ -D-glucopyranoside.

Several attempts to purify the second novel compound DE-O-Glcp did not yield sufficient quantities to allow acquisition of NMR data. However, the high-resolution mass spectrum of DE-O-Glcp suggested a molecular formula of  $C_{21}H_{20}O_{10}$  ( $[M+H]^+$   $m/z$  433.1129, calcd.  $C_{21}H_{21}O_{10}^+$  433.1129,  $\Delta M$  0.0 ppm);  $C_{21}H_{20}O_{10}$  ( $[M+H]^+$   $m/z$  431.0981, calcd.  $C_{21}H_{19}O_{10}^+$  431.0984,  $\Delta M$  0.7 ppm); and the loss of a hexose moiety ( $m/z$  162.0530) generated a fragment with the same mass as DE. These results indicate that this compound is an O-glucosylated form of DE, but the exact position of glucosylation could not be established.

The third novel compound was only detected in extracts from fresh and frozen *D. coccus* and identified as flavokermesic acid anthrone (FKA) (Fig. 2). The high-resolution mass spectrum of FKA detected in

the insect suggested a molecular formula of  $C_{16}H_{12}O_6$  ( $[M-H]^-$   $m/z$  299.0559, calcd.  $C_{16}H_{11}O_6^-$  299.0561,  $\Delta M$  1.0 ppm). Positive identification of this compound was achieved by comparison of retention time, high-resolution MS, MS/MS fragmentation pattern, and UV/VIS spectrum for FKA that had been prepared by chemical semi-synthesis. See Table S1 for NMR data. Chemical semi-synthesis of FKA demonstrated that the pure compound is prone to dimerization as well as oxidation to FK in the presence of oxidants under *in vitro* conditions. The observed spontaneous oxidation to FK likely explains why only fresh and frozen *D. coccus* was found to contain FKA while the compound was not detected in dried insects which have been exposed to light, oxygen and other oxidative agents for longer periods of time.

### 3.3. LC-DAD/MS-based screening of coccid dye production in selected members of the superfamily Coccoidea

To analyze the taxonomic distribution of the ability to produce coccid dyes, we collected representatives of four different Coccoidea families found in Denmark. The collected species included nymphal states of *Coccus hesperidum*, *Pseudococcus longispinus*, *Palmiculator browni* and *Pseudaulacaspis pentagona*. Metabolites from approximately 1 g of each of the individual species were extracted and analyzed by LC-MS/DAD. Extracted ion chromatograms (Fig. 3) for masses equivalent to the known coccid dye intermediates showed that *C. hesperidum* contained

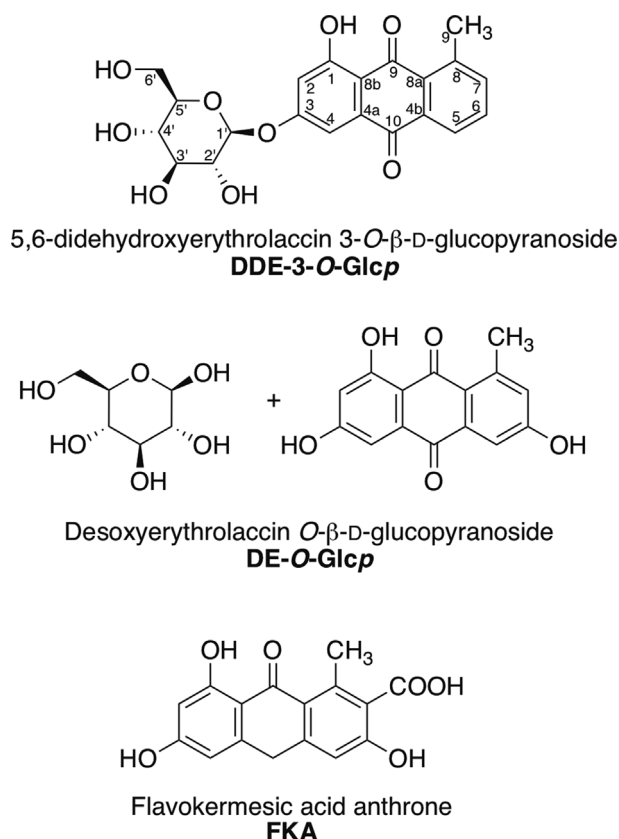


Fig. 2. Structures of the new compounds DDE-3-O-Glcp, and FKA.

FK, dcII and CA, while *P. longispinus* and *P. browni* both contained FK and dcII. Positive identification was based on accurate mass, retention time, UV/VIS spectra and authentic standards. Analysis of the *P. pentagona* material did not reveal the presence of any coccid dye intermediates in this species.

#### 4. Discussion

Assyrian cuneiform texts document that the utilization of scale insects for the production of pigments and dyes dates back to at least 3000 BCE (Dapson, 2007). Different civilizations separated by time and geography have relied on a limited number of scale insect species and the pigments they produce (Morgan, 2010; Cameron et al., 1978, 1981; Brown, 1975; Petitpierre, 1981; Pegg et al., 2007; Stathopoulou et al., 2013; Bhide et al., 1969). The long history and traditions associated with coccid dye utilization have resulted in a situation where scientific literature has focused on a very limited number of producing species (*Dactylopius coccus*, *Kerria lacca*, and *Kermes vermilio*) indicating a discontinuous taxonomic distribution of the trait and unrelated biosynthetic origins for the various pigments. However, the number of shared metabolites and the common FK core structure of the pigments suggest that all coccid dyes are formed via a similar biosynthetic scheme with a common evolutionary origin. If so, then many more scale insect species, i.e. those forming a monophyletic clade with known producers, would also be expected to be able to produce coccid dyes or related metabolites. To this end, we have demonstrated that members of the *Planococcus* (mealybugs) and *Pseudaulacaspis* genera also produce coccid dyes. Hence, the ability to form these pigments is indeed more widely taxonomically spread than previously believed. This would support the most parsimonious explanation, being a common evolutionary origin within Coccoidea. Of evolutionary importance here is that *Porphyrophora* and *Dactylopius* belong to two vastly different clades of Coccoidea. *Porphyrophora* belongs to the more primitive Margarodidae and

*Dactylopius* to the Dactylopiidae within the more highly derived Neococcoidea clade. The Margarodidae and Dactylopiidae share a most recent common ancestor 250 million years ago (Vea et al., 2016).

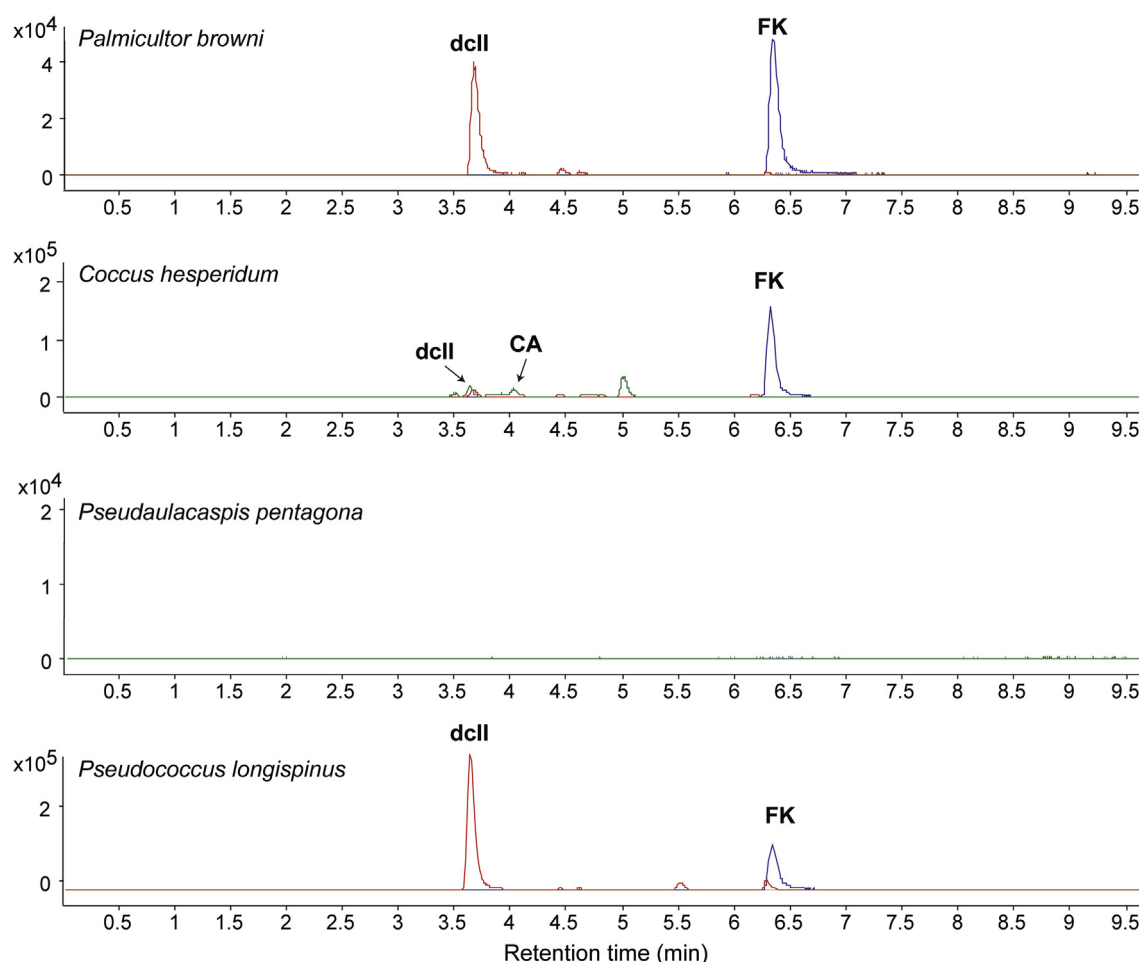
##### 4.1. The biosynthetic origin of coccid dyes

The biosynthetic origin of coccid dyes such as KA, FK, CA, and LA has long been debated (Brown, 1975; Joshi and Lambdin, 1996; Ramirez-Puebla et al., 2010). Several studies have rejected the hypothesis that host plants supply the insects with any of the known coccid dye intermediates. Similarly, we were unable to detect any of these intermediates in *Opuntia* cacti pads in the case of *D. coccus* (data not shown). To say nothing about the wide variety of different hosts utilized by many Coccoidea, e.g. over 400 plant species are described as host for *Kerria lacca* (Sharma et al., 1997).

Accordingly, it seems clear that Coccoidea must be able to synthesize the coccid dyes *de novo* from simple metabolites, e.g. glucose, present in the phloem sap of a wide range of plant species. Several studies have suggested that endosymbiotic bacteria may be responsible for the formation of the coccid dyes, and members of the *Wolbachia* and *Azoarcus* bacteria genera have been identified in the scale insects (Brown, 1975; Ramirez-Puebla et al., 2010; Pankewitz et al., 2007). However, none of these studies have proven a direct link between the presence, or activity, of these endosymbionts with the formation of coccid dyes. An alternative hypothesis is that the biosynthetic apparatus is encoded in the insects' nuclear genome. Several examples exist where complex secondary metabolite's biosynthetic pathways are encoded by genes in the genome of the producing insect, such as the *Drosophila* eye pigments drosoperin and ommochromes (Chovnick et al., 1990; Nijhout, 1997). Though no one has yet succeeded in identifying or describing the enzymological- or genetic basis for polyketide biosynthesis in insects, strong evidence does exist in support of nuclear encoding of the enzymes required for producing coccid dyes. However, this has largely been overlooked in past studies. Such evidence includes the report of a stable yellow color mutant of the normally red *Kerria lacca* (Indian Lac-insect) (Negi, 1954), and a white color mutant incapable of producing LAs (Chauhan and Teotia, 1973). Dissection of the genetic basis for these two mutations, by classical genetic crossing experiments, showed that the two traits are non-allelic and that they follow simple recessive inheritance (Chauhan, 1977; Chauhan and Mishra, 1977). If endosymbiotic bacteria were involved in catalyzing steps in the formation of the core structure of the coccid dyes, the mutant trait would be expected to be maternally inherited as a result of transfer via eggs (Ferrari and Vavre, 2011). Any conclusions on this matter must await demonstration of the ability of the color mutants to host endosymbiotic bacteria. Independently, support of an insect rather than bacterial origin of CA has been provided by a series of microscopy studies of the hemolymph from various *Dactylopius* spp. These studies described the existence of special granulocytes with a high concentration of secretory (M-) granules containing red pigments and small corpuscles of CA floating freely in the hemolymph (Joshi and Lambdin, 1996; Caselin-Castro et al., 2008, 2010).

##### 4.2. Model for the biosynthesis of carminic acid

Based on its structure, the biosynthesis of CA may be hypothesized to proceed by two different routes (Fig. 4). One envisioned biosynthetic scheme involves the polyketide pathway, also known as the acetate/malonate pathway (Fig. 4A). A second possible biosynthetic route involves the shikimate based chorismate/O-succinyl benzoic acid pathways (Fig. 4B). Both biosynthetic schemes ultimately result in the formation of anthraquinones; however they would be predicted to include different intermediates, e.g. a unique anthrone in the case of a polyketide-based pathway, which can be used to distinguish between the two. Detection of the FK anthrone (FKA) in fresh and frozen *D. coccus* material (Fig. 1) provides support for a polyketide rather than



**Fig. 3.** Targeted metabolite analysis of coccid dyes in representatives of the *Coccus*, *Pseudococcus*, *Hemiberlesia*, and *Palmiculor* genera. Extracted ion chromatograms at  $m/z$  313.0351  $\pm$  0.01 equal to the  $[M-H]^+$  ion of FK (green), at 475.0875  $\pm$  0.01 for the  $[M-H]^+$  ion of dclI (red), and at 491.0832  $\pm$  0.01 for the  $[M-H]^+$  ion of CA (green).

shikimate origin of compounds with the FK core. That the anthrone is detected in the current study can likely be attributed to the milder extraction conditions, and the use of fresh material rather than dried insects or dyed textiles that have been used in previous reports on the subject. The anthrone is abundant in the fresh material as evident by the HPLC-DAD-HRMS analysis (Fig. 1), but is not detected in dried insects, which is in good agreement with our previous observation that the purified anthrone oxidizes spontaneously. Microbial based reduction of the FK anthraquinone to yield the FKA anthrone is a possible alternative explanation for detection of the anthrone. de Witte and co-workers have previously shown that bacteria isolated from mammalian fecal material are capable of catalyzing the reduction of the anthraquinone rhein to the corresponding anthrone. The currently available data do not allow us to rule out this explanation, and further experiments e.g. feeding experiments are hence required (de Witte et al., 1992).

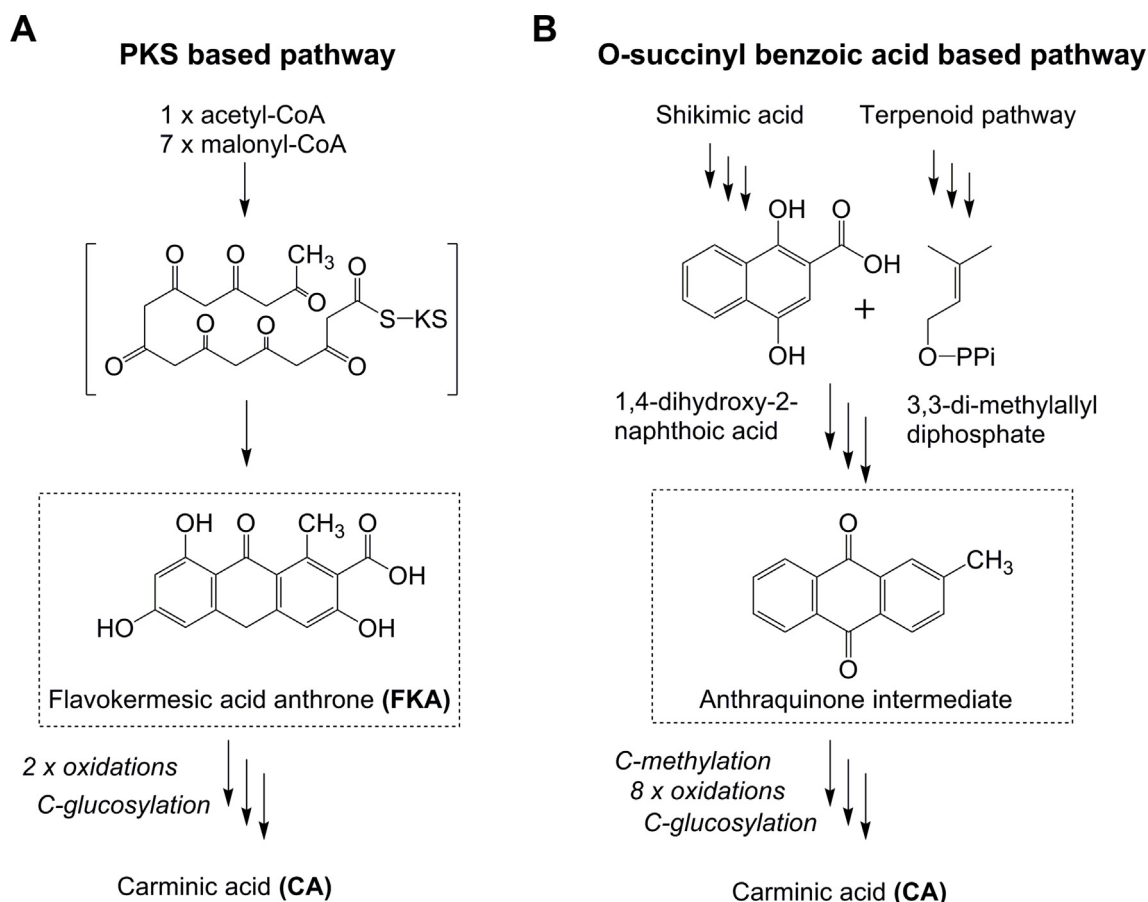
Based on the detection of FKA and the novel CA related compounds identified in *D. coccus*, and under the assumption that its formation is not the result of microbial reduction, we here propose a biosynthetic pathway for the formation of CA in *D. coccus* (Fig. 5). This biosynthetic pathway is an elaboration of the models previous proposed by Brown (1975) and Morgan (2010), and differs by including additional intermediates and predictions for the required enzymatic activities and co-factors.

The enzymatic machinery, responsible for the formation of polyketides in animals, remains unknown, and several competing hypotheses exists. One possible explanation could be that the involved

PKS has been introduced into the genome of scale insects by horizontal gene transfer (HGT) from fungi or bacteria. Several examples of HGT from fungi to insects have previously been documented such as the carotenoid forming pathway in *Acyrtosiphon pisum* (pea aphid) (Moran and Jarvik, 2010). Synthesis of the FK core requires the formation of a C7-C12 intermolecular bond in the octaketide backbone (Fig. 5). Since fungal type I iterative PKSs have only been described to form either C2-C7 or C6-C11 bonds, fungi are an unlikely donor (Li et al., 2010). Engineered bacterial type II PKS systems have previously been shown to be able to produce FK, known as TMAC in the bacterial literature (Tang et al., 2004). Specifically, TMAC is formed by the combined actions of the minimal actinorhodin PKS (act-KS $\alpha$ , act-KS $\beta$ , act-ACP) from *Streptomyces coelicolor* and the two cyclases (ZhuI and ZhuJ) from *Streptomyces* sp. No. 1128 (Tang et al., 2004). To settle whether HGT has formed the basis for CA production requires that the responsible genes are identified and analyzed in the context of a high-quality scale insect genome sequence, which is not presently available.

A competing hypothesis for the origin of insect PKSs, and the one we favor, is that the putative PKSs may have evolved from the insect's endogenous type I fatty acid synthase (FAS). Animal type I FAS and fungal type I iterative PKS are thought to have evolved from a common bacterial type I PKS ancestor (Hopwood and Sherman, 1990; Kroken et al., 2003). The main product of FASs in animals is palmitic acid, a fully reduced C-16 chain. This chain length is equivalent to an octaketide, which is the intermediate required for FKA formation. Converting a FAS to a non-reducing PKS capable of producing a non-reduced linear octaketide would require inactivation of the FAS's  $\beta$ -





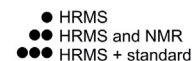
**Fig. 4.** The two theoretical biosynthetic schemes that can lead to the formation of carminic acid with the first stable tricyclic intermediates shown in boxes. Panel A shows the polyketide based pathway while B shows the O-succinyl benzoic based mechanism.

ketoreductase (KR) domain combined with a relaxation of the substrate specificity of its  $\beta$ -ketosynthase domain (KS) to allow for non-reduced products to form. Non-reduced linear polyketides are highly reactive, due to the presence of carbonyl groups on every second carbon atom, and they spontaneously fold into heterocyclic and aromatic structures via the formation of intramolecular C-C bonds. Several studies have shown that non-reduced octaketides spontaneously form the aromatic compounds **SEK4** and **SEK4b**, which contain skeletons that differ significantly from that of **FKA** (Fig. 5) (Fu et al., 1994; Mizuuchi et al., 2009). The fact that we find only **FKA**, and not **SEK4** or **SEK4b**, in *D. coccus* extracts (data not shown) suggests that folding of the polyketide chain does not proceed as a spontaneous reaction. Folding control of non-reduced polyketide backbones in fungal type I iterative non-reducing PKS systems and bacterial type III PKS systems are achieved by a 'Product Template' (PT) domain in the PKS (evolved from DH domain) or by *trans*-acting cyclases and aromatases, respectively (Shen et al., 1995; Bringmann et al., 2006). In the case of coccid pigments, controlled folding of the linear octaketide to form **FKA** could depend on similar mechanisms that would require additional mutations in the FAS or *trans*-acting enzymes as presented in Fig. 5. It has not escaped our notice that the mutated FAS hypothesis potentially also can explain the formation of other polyketides found in insects, such as 5-hydroxy-7-methyl-6-acetylpuropurin from *Ericoccus* spp. (Coccoidea: Ericoccidae) (Banks and Cameron, 1970), chrysophanol in *Galeruca tanacetii* (Coleoptera: Chrysomelidae) (leaf beetle) (Bringmann et al., 2006), and the predicted monomeric precursors of protoaphins in aphids (Brown, 1975). The listed compounds are all likely also formed from non-reduced octaketide precursors but display alternative backbone folds and would hence depend on other cyclases than those involved in coccid dyes biosynthesis.

In the case of **CA** formation, the enzymatic steps following formation of **FKA** are predicted to include two oxidations and a C-glucosylation. Based on the metabolites detected in *D. coccus*, monooxygenation of the central aromatic ring (position C-10), from **FKA** to **FK**, likely occur before oxidation of the outer ring (position C-4) and before C-glucosylation (position C-2), as neither of the detected metabolites contain a C-4 oxidation without a C-10 oxidation and as all known glucosylated intermediates (e.g. **dcII** and **CA**) have the C-10 oxidation. Several studies of bacterial and fungal systems have shown that efficient *in vivo* anthrone oxidation is dependent on specific anthrone oxidases (Chung et al., 2002; Ehrlich et al., 2010), making it likely that **FKA** to **FK** conversion is an enzyme dependent reaction *in vivo*, rather than a spontaneous reaction. The order of the two subsequent reactions (monooxygenation of C-4 and C-glucosylation at C-2) that ultimately yield **CA** is unclear as both **KA** and **dcII** accumulates. Hence, it is impossible to determine whether both pathways are active *in vivo* or whether one represents a shunt. Monooxygenation of the C-4 position that converts **FK** to **KA** is likely catalyzed by either a cytochrome P-450 or flavin-dependent monooxygenases. The accumulation of several intermediates suggests that the natural **CA** biosynthetic pathway is imbalanced; a situation that likely is caused by insufficient flux through downstream enzymatic steps resulting in the buildup of intermediates.

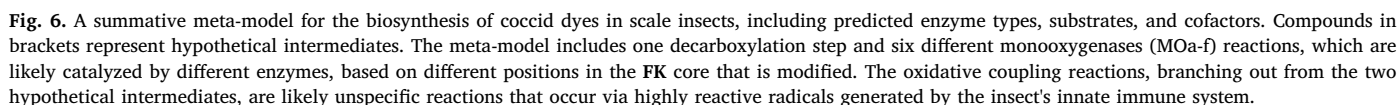
#### 4.3. Decarboxylation of the FKA core

We also demonstrated the presence of **DE**, **DE-O-GlcP**, and **DDE-3-O-GlcP** in freshly collected insects. **DE** has previously been observed in air-dried *D. coccus* (Sugimoto et al., 1998). Dehydroxy- and O-glucosylated forms of **DE** had not previously been reported in *D. coccus*. The reason why we detect the O-glucosylated forms may be attributed to the



mild extraction conditions and moderate pH compared to previously reported extraction protocols, which may result in hydrolysis of O-glucosides. *Kerria laccas* (lac insect) and *Austrotachardia acacia* (Maskell) are also known to accumulate DE, erythrolaccin (EL) and iso-erythrolaccin (IEL) in their resin deposits (Chauhan, 1977; Caselín-Castro et al., 2010). The structural similarity and co-occurrence in multiple species suggest a common biosynthetic origin for the FK (C16) and EL (C15) compound families. This is further supported by the observation made by Chauhan and Mishra (1977) who noted that a single mutation in white *K. lacca* strains affected both the body color (primarily caused by LA) and the resin color (primarily caused by EL) (Bhide et al., 1969).

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CA has been shown to act as a chemo deterrent that protects the immobile scale insects from predatory ants (Eisner et al., 1980). However, García Gil de Muñoz and co-workers recently extended CA's biological function by showing that it may contribute to the innate immune system of the scale insect to protect against invading microorganisms (García-Gil De Muñoz et al., 2002; 2005; De La Cruz Hernandez-Hernandez et al., 2003; García-Gil De Muñoz, 2007).

Specifically, this system depends on encapsulation of the invaders by melanization via the rapid formation of eumelanin by polymerization of tyrosine and L-DOPA (Satyavathi et al., 2014; Charles and Killian, 2015). Phenoloxidase (PO) is responsible for catalyzing multiple steps of the melanization cascade: tyrosine to L-DOPA, L-DOPA to dopaquinone, dopamine to dopaminequinone, and dopamine to *N*-arachidonoyl dopamine. Reactions that generate reactive radicals, including reactive oxygen species (ROS) and DOPA semi-quinones, which in addition to encapsulation may harm intruders (González-Santoyo and Córdoba-Aguilar, 2012). As shown by Garcia et al., PO can also act directly on

CA resulting in the formation of insoluble polymers of CA (García-Gil De Muñoz et al., 2005; García-Gil De Muñoz, 2007). This process is attributed to the quinone nature of CA allowing it to participate in redox cycling with ROSS. Relocation of the radical within the conjugated system of CA or FK would allow for activation of multiple positions in the FK core (C4, C8, C10, and C9-OH) as described for other naphthoquinones (Frandsen et al., 2006). This is a very interesting observation as the formation of FK radicals potentially can explain how the tyrosine-derived groups found in LAs may be added to the FK core by radical activation of the C10 positions and oxidative coupling with tyrosine or one of its derivatives (tyrosol, N-acetyltyramine, tyramine or 2-(4-hydroxyphenyl)ethyl acetate) (Fig. 6). Fig. 6 summarizes the chemical diversity and the required enzymatic steps in the form of a meta-biosynthetic pathway accounting for all known coccoid pigments. The model includes five different monooxygenases, acting on C4, C6, C8, C16, two dehydrogenases, a decarboxylase and a C-glucosyltransferase.

The extensive chemical diversity and existence of multiple alternative decoration patterns support the hypothesis that the FK forming biosynthetic pathway has a long evolutionary history within the *Coccoidea* superfamily.

In summary, we propose that formation of CA depends on the activity of a modified fatty acid synthase or polyketide synthase, possibly one or more cyclases/aromatases, one anthrone oxidases, a 'cytochrome P450 monooxygenases'/'flavin-dependent monooxygenases', and a C-glucosyltransferase. Validation of the proposed hypothetical biosynthetic schemes and the involved enzyme types naturally depends on future biochemical evidence and mapping of their genetic basis in *D. coccus* or an endosymbiont organism.

## Conflicts of interest

Authors SAR, KTK, DS, CHG, UT, UHM, TOL, and RJNF declare no financial nor any competing financial interests. The authors, PKJ, MB, BM, RMK, MN and FO are or were employed by the private company Chr. Hansen A/S that produces and sells *D. coccus* derived carmine as a food pigment in a business-to-business setup.

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## Appendix A. Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.ibmb.2018.03.002>.

## Abbreviations

FKA	Flavokermesic acid anthrone
FK	flavokermesic acid
KA	kermesic acid
CA	carminic acid
dcII	C-glucosylated flavokermesic acid
LA	laccic acid
DE	desoxyerythrolaccin

DDE	5,6-didehydroxyerythrolaccin
DDE-3-O-GlcP	didehydroxyerythrolaccin 3-O-β-D-glucopyranoside
EL	erythrolaccin
IEL	iso-erythrolaccin

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