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Feasibility of synthetic fuels in renewable energy systems



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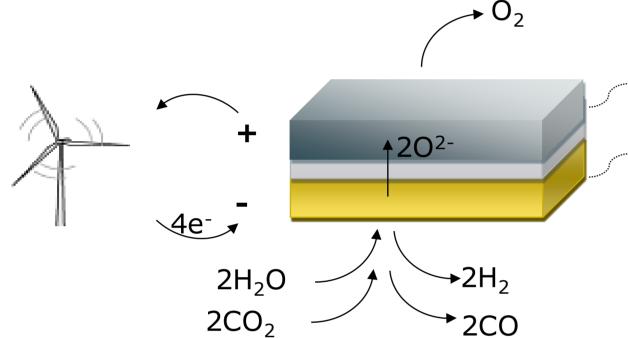
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The transport sector is the only sector in which there have been no significant renewable energy penetrations and it is heavily dependent on oil with rapid growth in the last decades. Moreover, it is challenging to obviate the oil dependence due to the wide variety of modes and needs in the sector. Nowadays, biofuels are proposed as one of the main options for replacing fossil fuels in the transport sector, along with electricity. The main reasons for avoiding the direct usage of biomass in the transport sector, i.e. producing biomass derived fuels, are land use shortage, limited biomass availability, interference with food supplies, and other impacts on environment and biosphere. Hence, it is essential to do a detailed analysis of the transport sector in order to match the demand and to meet the criteria of a 100% renewable energy system in 2050.

SOLID OXIDE ELECTROLYSER CELLS

The advantages of solid oxide electrolyser cells are the potential for great fuel production rates at high efficiency, low material costs and the possibility of co-electrolysis of H_2O and CO_2 . Solid oxide electrolyte conducts oxide ions which cannot be done with other types of cell. High operating temperature and high pressure, which provides further efficiency improvement, enables the integration of catalysis of the synthetic gas to synthetic fuel.

The main disadvantage of SOECs is the durability of the cell - durable performances at high current densities remain to be proven.



PRODUCTION CYCLE OF SYNTHETIC FUELS

Electricity from

Carbon and

energy source

Porous positive electrode (anode) $20^{2-} \rightarrow 0_2 + 4e^{-}$ Porous negative electrode (cathode) $2H_2O + 4e^- \rightarrow 2H_2 + 2O^{2-}$ $2CO_2 + 4e^- \rightarrow 2CO + 2O^{2-}$ Net cell reactions $2H_2O + electricity \rightarrow 2H_2+O_2$ $2CO_2$ + electricity \rightarrow $2CO+O_2$

mixtures for

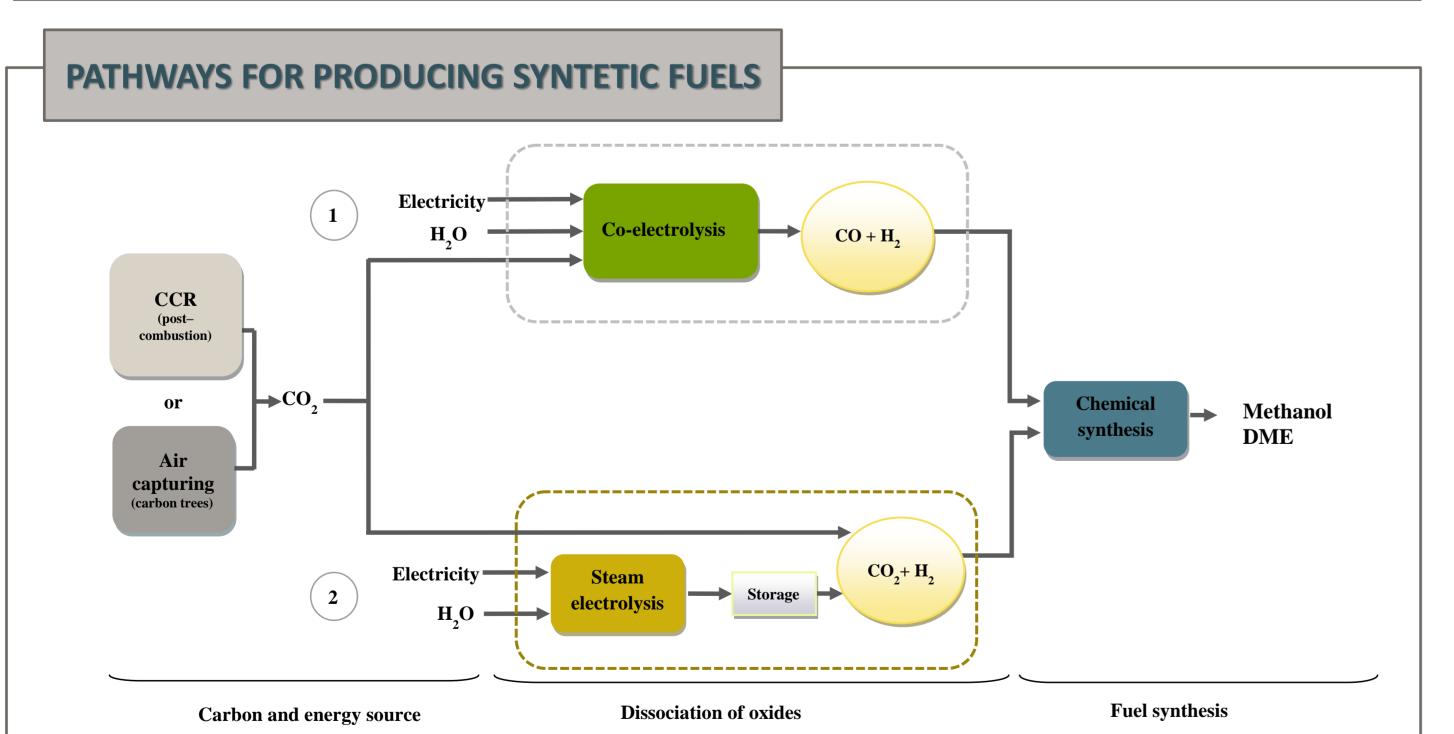
production of synthetic fuels

Fuel synthesis

Dissociation of oxides CCS or Air Chemical Co-electrolysis synthesis of gas capturing Steam

electrolysis

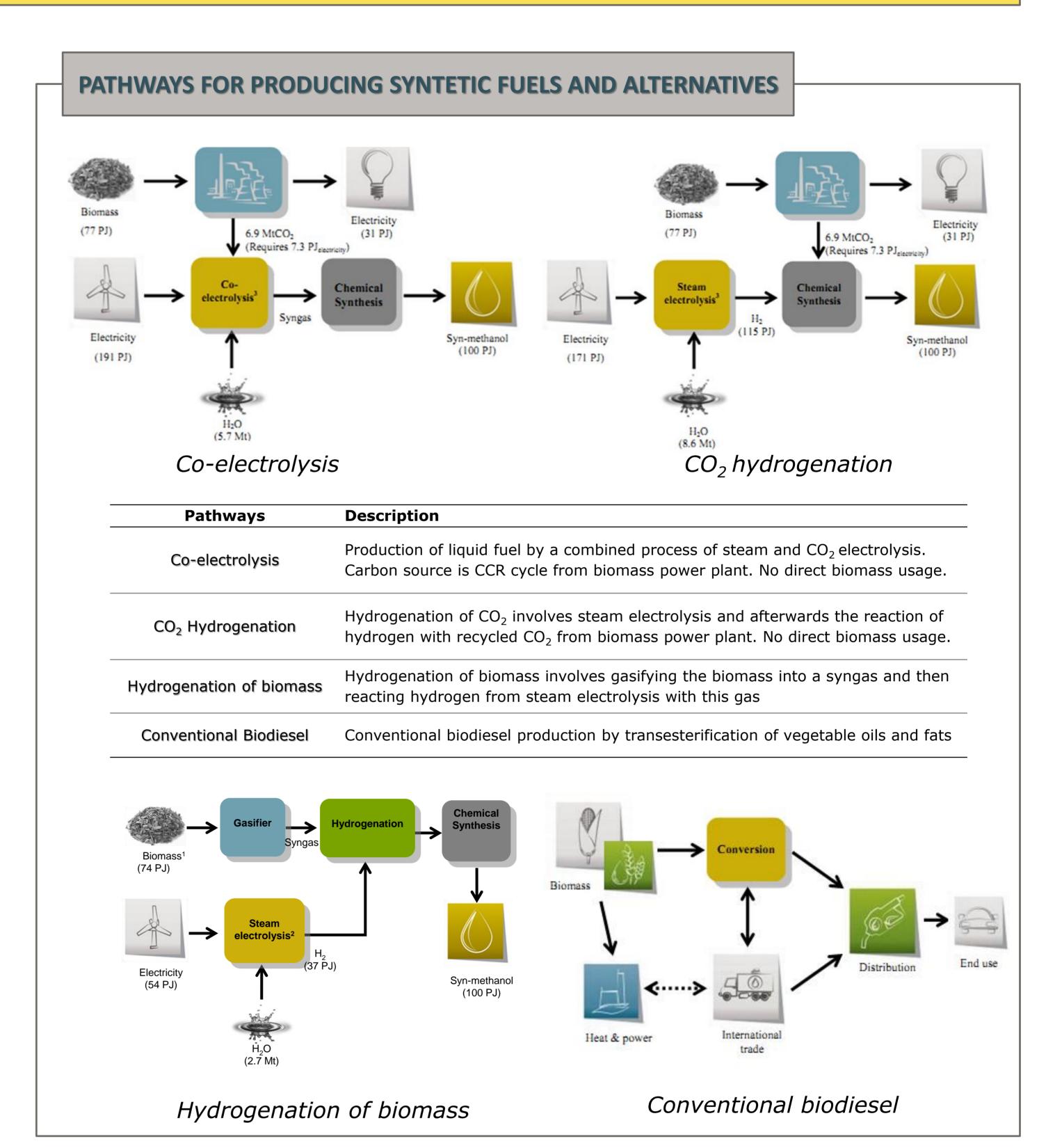
Two carbon sources are proposed - CCR from energy sector or air capturing as a promising future technology. Pathways are divided into three steps. Energy source drives dissociation of oxides, either H₂O or both H₂O and CO₂, which results in gas mixtures that are in the last step catalyzed to fuels.

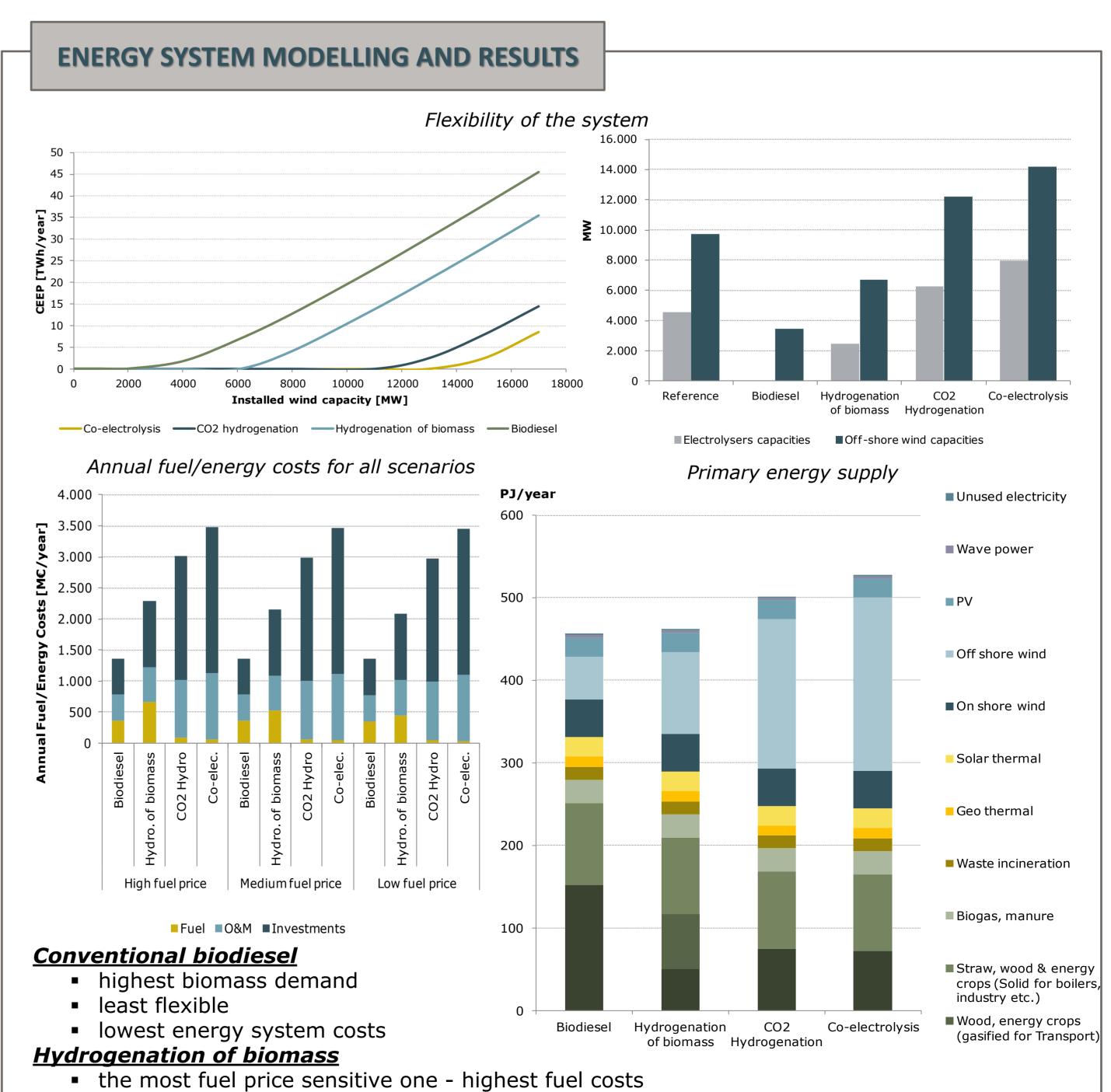


Both pathways for producing synthetic fuels exclude direct biomass input for fuel production. However, these pathways are in strong connection with power and heat sector that uses biomass

CONCLUSION

Costs of synthetic fuel scenarios are more expensive, but biomass savings associated with this make the additional costs worthwhile due to the scarcity of biomass for the energy system. With feasible technological development and mass production of the Solid Oxide Electrolyser Cells, synthetic fuels could be competitive and have market advantage over biomass derived fuels based on their supply related issues, land use shortage, limited biomass resources, etc.





large investment costs - transport demand was met by electricity from wind turbines and

Co-electrolysis and **CO₂ hydrogenation**

lowest fuel consumption

electrolysers

most flexible alternatives, high share of wind energy